Safety aspects of nanoparticle reactions and explosions, their detection/identification and protective/preventive measures

NANEX

merging with NANOWORKS

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Background

Nanoparticles are going to be used in larger scales:

Powders and porous structures in manufacturing and storage

Suspensions/Emulsions of nano-particles for storage and handling

Spraying and thin films for coatings and feeding

Inert atmospheres needed => high costs

Incorporated in materials with high surface contact

Guideline and regulation needed for industrial users – especially SMEs have not the means for investigations

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Effects

Explosion: severe explosions, faster conversion, higher pressure rises, Deflagration-Detonation-Tranfer enhanced, severe explosion of porous Si

Ignition: sensitivity against stimuli increased, temperatures lowered (> 1000 K for super thermites – metal/metaloxide mixtures)

<u>Thermal explosion in air:</u> critical diameters as for other hazardous materials: particles in air are oxidised at lower temperatures

<u>Surface reactions dominate at small diameters</u>: nanoscale reactions (e.g. passivation) become important at small diameters also with small reactions constants

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<u>Stability</u>:Long term compatibility of materials changes Fraunhofer In

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Scientific and technical objectives

Physical and chemical <u>characterisation</u> of nanoparticles and suspensions/sprays (with nanoparticles) and <u>measurement techniques</u> to detect and identification and assessment of hazards

Mechanisms:

Chemical reaction kinetics: single particle reactions, catalytic effects and cloud propagation, surface reactions, heterogeneous reactions

Phenomena: Ignition, self heating/thermal explosion, fire and explosions in nanoparticles processes, long term reaction on storage - compatibility

Simulation, prevention and protection against these hazards

Regulation and technical recommendations on handling nanoscale particles, suspensions and sprays

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Workprogramme

Measurement technologies

- Online/ in-situ measurement: detection, identification and kinetics to avoid risks
- methods for particle size, surface morphology, concentration, chemical composition
- easy to handle and on-/-inline applicable equipment including calibration and verification

• nanoparticles and suspensions/sprays (with nanoparticles) with respect to reactive properties, surface morphology (increased surface area compared to spheres, surface layers (oxide layers))

Basic reaction mechanisms and experiments

• Chemical reaction kinetics, diffusion of gases and liquids in nanostructures and surfaces, single particle reactions and cloud propagation reactions, solid-solid reactions, catalytic effects

- Slow long term reactions, stability and compatibility with other materials, accelerated measurements (T, P) Self heating/thermal explosion, ignition sources, fires and explosions of nanoparticles, nano-dusts and nano-structured materials
- Fires and explosions of nanoparticles, solids in liquids and liquid sprays, clouds

Verification; preventive/protective measures

- Verification by larger scale experiments, long time storage experiments, simulated by accelerating tests
- Simulation, prevention and protection against fire and explosions of nanoparticles and nanostructured materials (exposed to reactive gases)
- Definition test methods for various hazards which might later become standard
- Risk evaluation tools and regulations,

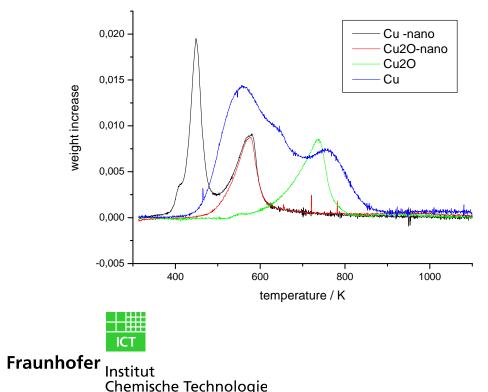
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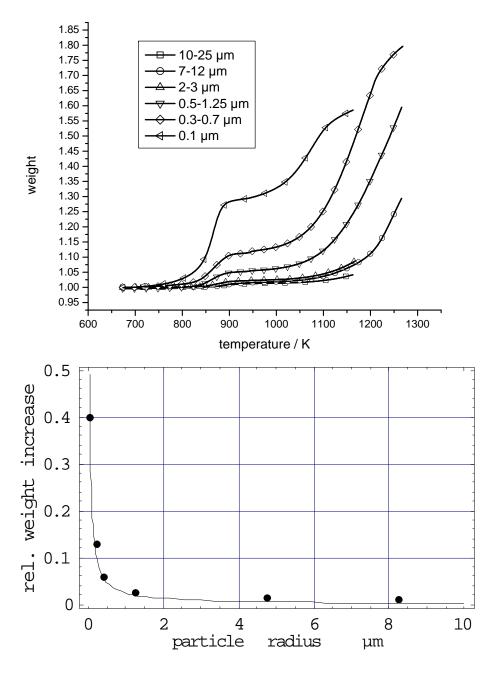
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Example: Oxidation behaviour of nano metal particles

nano-particles react:

- at lower temperatures
- in a different way, chemically dominated not only diffusion controlled





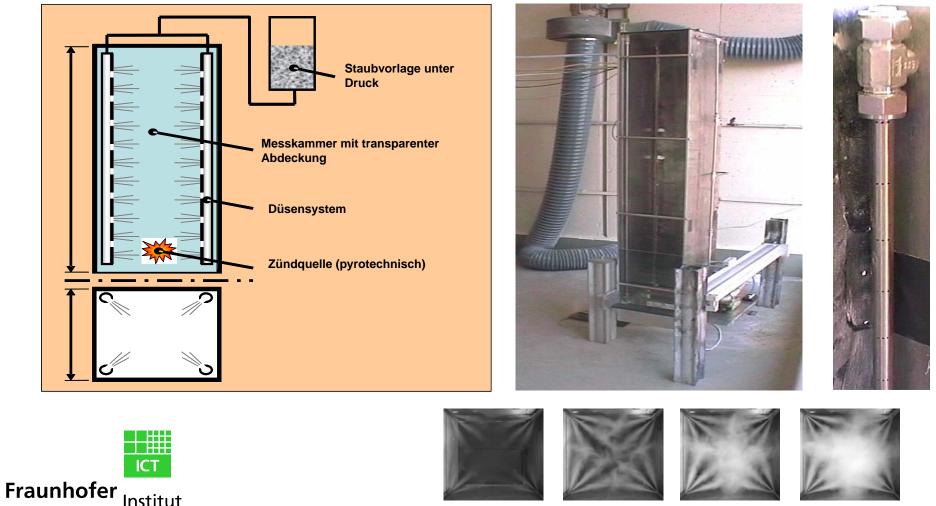
Al-powders oxidised at rate 5 K/min,

Al-slurry explosion in air





Fire/Explosion chamber: particles/sprays controlled turbulence Flame propagation depending on parameters Temperature/species profiles of flames and particles



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107 ms

113 ms

119 ms

125 ms